

Sequential Ruthenium-Catalyzed Hydroamination and Rhenium-Catalyzed C–H Bond Activation Leading to Indene Derivatives

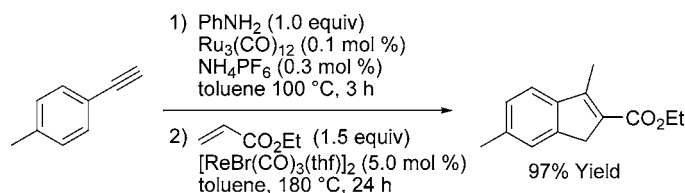
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ABSTRACT



Formal [3+2] annulation of arylacetylenes and α,β -unsaturated carbonyl compounds is achieved in a one-pot reaction by successive treatment of the acetylenes with aniline and a catalytic amount of $\text{Ru}_3(\text{CO})_{12}$ and NH_4PF_6 and C–H bond activation catalyzed by $[\text{ReBr}(\text{CO})_3(\text{thf})_2]$. The result suggests that the second rhenium-catalyzed indene formation is not disturbed by the first catalyst system.

Recently, one-pot sequential reactions using two or more metal catalysts have received considerable attention as efficient chemical transformations because they estimate a number of workup steps, which are bothersome and often cause a decrease in yields.¹ We have previously reported on the rhenium-catalyzed synthesis of indene derivatives, which are partial structures in functional materials,² medicines,³ and organic compounds,⁴ via C–H bond activation. We achieved this using aromatic ketimines or ketimines generated in situ by reactions between ketones and anilines.⁵ The other route to prepare ketimines is hydroamination of aromatic acetylenes with amines (Figure 1).⁶ In this study, we examined the successive hydroamination of aromatic acetylenes and

rhenium-catalyzed C–H bond activation of the formed aromatic ketimines leading to indene derivatives.

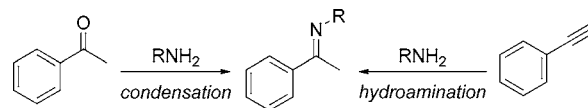


Figure 1. Preparation methods of aromatic ketimines.

Initially, we attempted to use Cp_2TiMe_2 ⁷ and AuCl_3 ⁸ for the hydroamination step. Although an aromatic ketimine was generated from phenylacetylene and aniline with the catalysts, the corresponding indene derivative was not obtained

(1) Shibasaki, M.; Yamamoto, Y., Eds. *Multimetallic Catalysts in Organic Synthesis*; Wiley-VCH: Weinheim, 2004.

(2) (a) Akbulut, U.; Khurshid, A.; Hacıoglu, B.; Toppare, L. *Polymer* **1990**, *31*, 1343. (b) Barbera, J.; Rakitin, O. A.; Ros, M. B.; Torroba, T. *Angew. Chem., Int. Ed.* **1998**, *37*, 296. (c) Yang, J.; Lakshminathan, M. V.; Cava, M. P. *J. Org. Chem.* **2000**, *65*, 6739.

(3) Korte, A.; Legros, J.; Bolm, C. *Synlett* **2004**, *13*, 2397 and references therein.

(4) Miller, R. B.; Frincke, J. M. *J. Org. Chem.* **1980**, *45*, 5312.

(5) Kuninobu, Y.; Nishina, Y.; Shouho, M.; Takai, K. *Angew. Chem., Int. Ed.* **2006**, *45*, 2766.

(6) (a) Pohlki, F.; Doye, S. *Chem. Soc. Rev.* **2003**, *32*, 104. (b) Bytschkov, I.; Doye, S. *Eur. J. Org. Chem.* **2003**, 935.

(7) For the titanium complex-catalyzed hydroamination of acetylene with aniline, see: Haak, E.; Bytschkov, I.; Doye, S. *Angew. Chem., Int. Ed.* **1999**, *38*, 3389.

(8) Luo, Y.; Li, Z.; Li, C.-J. *Org. Lett.* **2005**, *7*, 2675.

by successive treatment of the ketimine with a rhenium complex, $[\text{ReBr}(\text{CO})_3(\text{thf})_2]$, in toluene at 180 °C for 24 h. On the other hand, when a ruthenium complex, $\text{Ru}_3(\text{CO})_{12}$,⁹ was used as the catalyst for hydroamination, the sequential reactions proceeded as we expected. We report herein the ruthenium- and rhenium-catalyzed one-pot synthesis of indene derivatives by the reactions of aromatic acetylenes with aniline followed by treatment with α,β -unsaturated carbonyl compounds.

Treatment of phenylacetylene **1a** with aniline and a catalytic amount of $\text{Ru}_3(\text{CO})_{12}$ and NH_4PF_6 at 100 °C in toluene generated an aromatic ketimine,^{9,10} which was then treated with ethyl acrylate **2a** in the presence of $[\text{ReBr}(\text{CO})_3(\text{thf})_2]$, in toluene at 180 °C for 24 h. The indene derivative **3a** was obtained via ruthenium-catalyzed hydroamination and rhenium-catalyzed C–H bond activation in 82% yield (Table 1, entry 1).^{11,12} This result indicates that the first-step conditions such as the choice of catalysts and solvents do not disturb the second reaction with the rhenium catalyst.

Acetylenes having electron-donating groups at the para position of the aromatic ring, **1b** and **1c**, gave the corresponding indene derivatives **3b** and **3c** in good to excellent yields, respectively (Table 1, entries 2 and 3). However, an indene derivative was not formed when using an acetylene bearing an electron-withdrawing group at the para position of the aromatic ring.¹³ An acetylene having a methyl group at the ortho position of the aromatic ring gave the corresponding indene derivative **3d** in 26% yield (Table 1, entry 4). To investigate the selectivity, nonsymmetrical acetylene **1e** was used. As a result, indene derivatives **3e** and **4e** were obtained as a mixture of isomers (Table 1, entry 5). 2-Ethynyl-naphthalene also gave a mixture of two isomers, **3f** and **4f**, in 81% combined yields (Table 1, entry 6).

Next, we investigated the substituent of the α,β -unsaturated esters. By treatment of phenylacetylene **1a**, aniline, and phenyl acrylate **2b**, the corresponding indene derivative **3g** was formed in moderate yield (Table 1, entry 7). Vinyl ketone **2c** also generated an indene derivative **3h** in moderate yield (Table 1, entry 8), although the corresponding indene derivatives were not obtained when using styrene, acrylonitrile, and *tert*-butyl acrylate, due to polymerization.

Because phenyl acetylenes polymerize gradually with the rhenium complex, the yield of **3a** decreased to 21% when both ruthenium and rhenium catalysts were added with aniline at the same time.¹⁴

We consider the proposed reaction mechanism as follows:¹⁵ (1) ruthenium-catalyzed hydroamination of an aro-

Table 1. Reactions of Aromatic Acetylenes with α,β -Unsaturated Esters

entry	R	R'	product	% yield ^{a,b}
1	1a (R = H)	2a (R' = OEt)	3a	82 (85)
2	1b (R = MeO)	2a	3b	63 (75)
3	1c (R = Me)	2a	3c	97 (>99)
4	1d	2a	3d	26 (32)
5	1e	2a	3e 4e	72 (80) [82:18]
6	1f	2a	3f 4f	81 (87) [77:23]
7	1a	2b (R' = OPh)	3g	66 (72)
8	1a	2c (R' = Et)	3h	49 (57)

^a Isolated yield. The yield determined by ¹H NMR is reported in parentheses. ^b The ratio between **3** and **4** is reported in brackets.

matic acetylene with aniline; (2) rhenium-catalyzed C–H bond activation at the ortho position of an aromatic ketimine;^{5,16} (3) insertion of an α,β -unsaturated carbonyl compound into a carbon–rhenium bond; (4) intramolecular nucleophilic cyclization; (5) reductive elimination and the elimination of aniline (or vice versa) (Scheme 1).

(15) (a) Neither the combination of the ruthenium complex, $\text{Ru}_3(\text{CO})_{12}$, NH_4PF_6 , and aniline nor the combination of the rhenium complex, $[\text{ReBr}(\text{CO})_3(\text{thf})_2]$, NH_4PF_6 , and aniline gave indene derivatives.

(16) (a) Kunitobu, Y.; Kawata, A.; Takai, K. *J. Am. Chem. Soc.* **2005**, *127*, 13498. (b) Kunitobu, Y.; Tokunaga, Y.; Kawata, A.; Takai, K. *J. Am. Chem. Soc.* **2006**, *128*, 202.

(9) Tokunaga, M.; Eckert, M.; Wakatsuki, Y. *Angew. Chem., Int. Ed.* **1999**, *38*, 3222.

(10) Because hydroamination of internal acetylenes does not proceed with $\text{Ru}_3(\text{CO})_{12}$,⁹ the procedure is applicable for terminal alkynes.

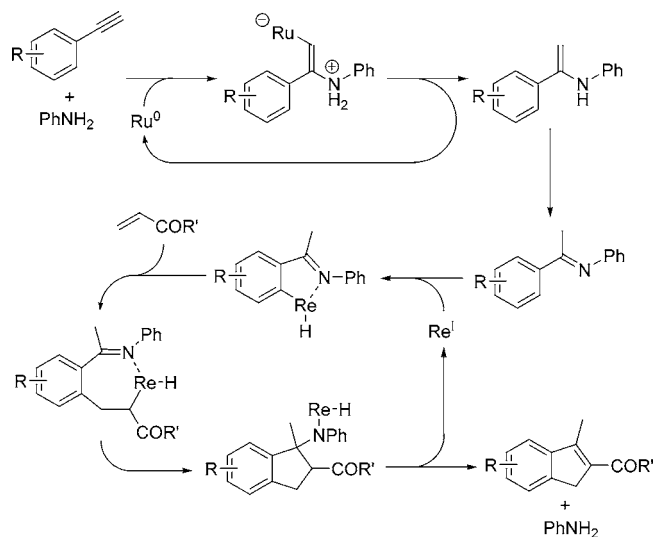
(11) The ¹H NMR yields were determined by using 1,1,2,2-tetrachloroethane as an internal standard.

(12) *p*-Anisidine gave indene derivative **3a** in almost the same yield as aniline. However, cyclohexylamine did not provide **3a** at all.

(13) The electron-withdrawing group decreased the reactivity in both the ruthenium-catalyzed hydroamination and the rhenium-catalyzed formation of indene frameworks. This might be the main reason an indene derivative was not formed by using 1-ethynyl-4-trifluoromethylbenzene.

(14) Treatment of a mixture of **1a** and **2a** with 50 mol % of aniline in the presence of $\text{Ru}_3(\text{CO})_{12}$ (1.0 mol %), NH_4PF_6 (3.0 mol %), and $[\text{ReBr}(\text{CO})_3(\text{thf})_2]$ (3.0 mol %) at 180 °C for 24 h gave **3a** in 21% yield.

Scheme 1. Proposed Mechanism of the Formation of Indene Derivatives



In summary, we have succeeded in the synthesis of indene derivatives by the formal [3+2] annulation of aromatic

terminal acetylenes with α,β -unsaturated carbonyl compounds. These reactions contain two important reaction steps: (1) ruthenium-catalyzed hydroamination of an aromatic acetylene with aniline; (2) rhenium-catalyzed reaction of the aromatic ketimine with an α,β -unsaturated carbonyl compound via C-H bond activation, conjugate addition, intramolecular nucleophilic cyclization, and the elimination of aniline.

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Supporting Information Available: General experimental procedures, characterization data, and ¹H and ¹³C NMR spectra for indene derivatives (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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